

B .V.V.Sangha's  
Basaveshwar Science College, Bagalkot  
**DEPARTMENT OF PHYSICS**  
**Slow Learners Students list**  
**It Semester 2023-24**

Sl. No	Register No.	Name of Students
1	U26ZW23S0020	HULAGAPPAGOUDA NANDIHAL
2	U26ZW23S0023	SAMEER D. NADAF
3	U26ZW23S0081	MALLIKARJUN
4	U26ZW23S0141	RASHMI SALAPUR
5	U26ZW23S0213	PAVAN GARAWAD
6	U26ZW23S0216	HARISH BILLUR
7	U26ZW23S0219	BASAVARAJ R. NAGAPPANAVAR
8	U26ZW23S0254	NIRIKSHA HADIMANI
9	U26ZW23S0347	SUNIL S. HULYAL
10	U26ZW23S0379	VINODKUMAR ARIKERI
11	U26ZW23S0385	OMAR ABDULLA
12	U26ZW23S0419	PARVATI D. UPPAR
13	U26ZW23S0444	NISARGA MURADI
14	U26ZW23S0461	SURAJ

**H.O.D**

Department Of Physics  
B.V.V.S.'s Basaveshwar  
Science College, Bagalkot

**PRINCIPAL**  
Basaveshwar Science College  
BAGALKOT



B .V.V.Sangha's  
Basaveshwar Science College, Bagalkot  
DEPARTMENT OF PHYSICS

Slow Learners Students list  
IV Semester 2023-24

Sl. No	Register No.	Name of Students
1	U15JP22S0025	Tejaswini Badiger
2	U15JP22S0077	Shweta Kanabor
3	U15JP22S0078	Vikas Chandakavate
4	U15JP22S0098	Amreen Kolhar
5	U15JP22S0159	Altaf Kakhandakai
6	U15JP22S0174	Vaibhav Jagajinni
7	U15JP22S0178	Yamanavva Mantur
8	U15JP22S0182	Tippanna Tumbaramatti
9	U15JP22S0216	Akhilesh Lamani
10	U15JP22S0286	Hanamant Bhajantri
11	U15JP22S0306	Sagar Ittagi
12	U15JP22S0369	Pratik Badiger
13	U15JP22S0370	Harish Talawar

**H.O.D**

Department Of Physics  
B.V.V.S.'s Basaveshwar  
Science College, Bagalkot

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BAGALKOT

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Basaveshwar Science College, Bagalkot  
DEPARTMENT OF PHYSICS



Slow Learners Students list  
VI Semester 2023-24

Sl. No	Register No.	Name of Students
1	U15JP2150013	Amisha S Palled
2	U15JP2150035	Arun H Malani
3	U15JP2150111	Chandrashekhar S Patil
4	U15JP2150210	Jyoti G Tolamatti
5	U15JP2150229	Kavita Javoor
6	U15JP2150234	M Swathi
7	U15JP2150245	Mallu Basappa Kunibenchi
8	U15JP2150273	Anuradha M Birdar
9	U15JP2150305	Muttuvinayak Walikar
10	U15JP2150307	Nagaraj S Hiremath
11	U15JP2150316	Panchakshrayya K Salimath
12	U15JP2150324	Rahul B Ganiger
13	U15JP2150327	Rahul M Pawar
14	U15JP2150341	Ramesh R Hangaragi
15	U15JP2150345	Rakshitha C Patil
16	U15JP2150362	Sandesh D Bandiwaddar
17	U15JP2150363	Sachin M
18	U15JP2150365	Sachin N Allagi
19	U15JP2150389	Shashidhar Oni
20	U15JP2150394	Shravana D Mayachari
21	U15JP2150414	Sushma I Havaragi
22	U15JP2150421	Simran M Batakuraki
23	U15JP2150425	Umesh Madar
24	U15JP2150433	Varasha A Angadi
25	U15JP2150435	Varshita R Dapalapup
26	U15JP2150454	Vijay P Gari
27	U15JP2150456	Vijayalaxmi Mahadevappa
28	U15JP2150468	Sandesh M A

**H.O.D**

Department Of Physics  
B.V.V.S.'s Basaveshwar  
Science College, Bagalkot


**PRINCIPAL**  
Basaveshwar Science College  
BAGALKOT

**B.V.V.Sangha's  
Basaveshwar Science College. Bagalkote  
Department of Chemistry  
B.Sc II semester**

**Student List ( Slow - Learners ) 2023 -2024**

SL No	Reg No	Name of the Student	Gender	Caste	Category
1	U26ZW23S0006	Sachin Irappa Kamashetti	Male	Lingayat	Category III (B)
2	U26ZW23S0029	VISHWANATH ASHOK LAMANI	Male	Lamani	Scheduled Caste
3	U26ZW23S0070	PRAVEEN KELUR	Male	Voddar	Scheduled Caste
4	U26ZW23S0078	Keerti Shrishail Karigar	Female	Kuruba	Category II (A)
5	U26ZW23S0119	RAMYA GURAPPANAVAR	Female	Valmiki	Scheduled Tribe
6	U26ZW23S0121	SHRIMANT PATIL	Male	Kurab	Category II (A)
7	U26ZW23S0137	MAHANTESH HANUMANTA KALLUK	Male	Waddar	Scheduled Caste
8	U26ZW23S0140	SHREENIDHI HIREMATH	Female	Veerashaiva	Category III (B)
9	U26ZW23S0164	Pradeep Hadimani	Male	Reddy	Category III (A)
10	U26ZW23S0167	KRISHNA S BHOSHANAVAR	Male	Kurab	Category II (A)
11	U26ZW23S0168	MANJUNATH V KELAGINONI	Male	Madar	Scheduled Caste
12	U26ZW23S0176	NOUSHAD MULLA	Male		Category II (B)
13	U26ZW23S0204	SRUSHTI MAREPPA BHOVI	Female	Bhovi	Scheduled Caste
14	U26ZW23S0215	ANUSHA	Female	Kuruba	Category II (A)
15	U26ZW23S0217	AKSHATA BHIMASHI ARAKERI	Female	Valmiki	Scheduled Tribe
16	U26ZW23S0219	BASAVARAJ RNAGAPPANAVAR	Male	Kurab	Category II (A)
17	U26ZW23S0254	NIRIKSHA HADIMANI	Female	Reddy	Category III (A)
18	U26ZW23S0260	CHETAN HUNAGUNDI	Male	Devang	Category II (A)
19	U26ZW23S0286	GANGADHAR B SEETIMANI	Male	Kurab	Category II (A)
20	U26ZW23S0299	POOJA PRAKASH MINAJAGI	Female	Navi	Category II (A)
21	U26ZW23S0347	SUNIL SHANKRAPP HULYAL	Male	Kuruba	Category II (A)
22	U26ZW23S0372	ANJALI SHIVANAND WADAWADAGI	Female	Devang	Category II (A)
23	U26ZW23S0385	OMAR ABDULLA	Male	Muslim	Category II (B)
24	U26ZW23S0389	MAHANTESH MALLAYYA MATH	Male	Lingayat	Category III (B)
25	U26ZW23S0397	LOKESHWARI SUNAGAD	Female	Valmiki	Scheduled Tribe
26	U26ZW23S0404	SUNIL HOKRANI	Male	Kurubar	Category II (A)
27	U26ZW23S0444	NISARGA MURADI	Female	Ganiga	Category II (A)
28	U26ZW23S0445	MALLIKARJUN SANGAPPA ROLLI	Male	K.Setty	Category II (A)
29	U26ZW23S0465	PRAJWAL B KALYANSHETTI	Male	Banajiga	Category II (A)

  
( Smt S.S.Karishankari )

  
**Principal**  
**B. V. V. S' s Basaveshwar**  
**Science College, Bagalkot**

**B.V.V.Sangha's  
Basaveshwar Science College, Bagalkote  
First Internal Assessment Test July-2024  
B.Sc-II Semester Slow learners**

**Sub: Chemistry**

**Date: / 06/2024**

**Marks :20 Marks**

**Time: 3.30 pm to 4.30 pm**

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**Q.No.I. Answer any THREE of the following.**

**2x4=08 Marks**

1. Define ionic bond.
2. Define Lattice energy
3. What is Geometrical isomerism? Give an example.
4. What is Stereoisomerism? Give the examples of stereoisomers.
5. Define hybridization. Mention its types.
6. Define order of a reaction.
7. Define viscosity coefficient

**Q.No.II. Answer any THREE of the following.**

**4x3=12Marks**

1. Discuss the shape of  $\text{BeF}_2$  molecule.
2. What are the steps involved in Born-Haber cycle.
3. Explain Newman and Sawhorse projection formulae of ethane.
4. Explain the confirmation analysis of n-Butane.
5. Explain the determination of surface tension of liquid by drop number method.
6. Explain the half life period method of determination of order of reaction

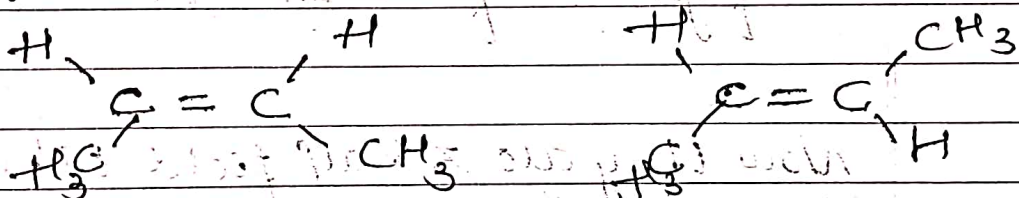
Q No. 1 Answer any 3 of the following

1] Ionic bond: The electrostatic force of attraction which holds the two oppositely charged ions together, is called the ionic bond.

2] Lattice Energy: The energy released when one mole of an ionic crystal is formed from one mole of gas +ve & one mole of gas -ve ions, is called the lattice energy.

3] Geometrical Isomers: The different geometrical arrangement of groups in a molecule which contains  $C=C$  groups are known as Geometrical Isomerism.

Ex:



Cis-2-butene and Trans-2-butene

[Cis Isomer] and [Trans Isomer]

7] Viscosity coefficient: The force per unit area required to maintain unit difference of velocity between 2 parallel layers separated by 1 cm apart is called viscosity.

1]  $\text{BF}_3$ : It is also the example of the  $sp^2$  hybridization.

One  $s$  orbital mixed with  $2p$  orbitals having different energy to form three new identical  $sp^2$  hybrid orbitals of equal energy.

2] Electron configuration of Boron in ground state is  $1s^2, 2s^2, 2p^1$ .

One of the two  $s$  electrons gets promoted to next available vacant orbital  $2p_y$ .

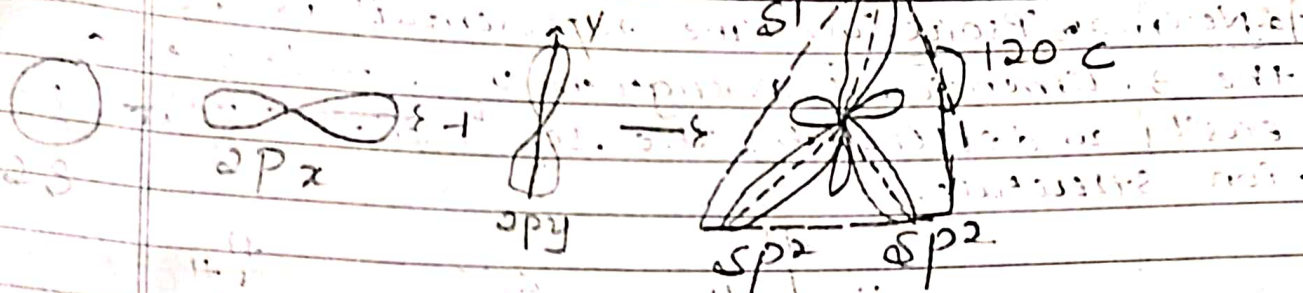
E.C. of boron in the excited state is  $1s^2, 2s^1, 2p^1, 2p^1$ .

Electron configuration of B atom in Ground state  $\uparrow\downarrow \uparrow\downarrow \uparrow$

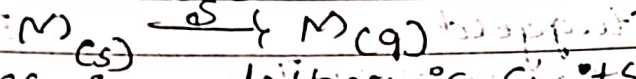
Electron configuration of B atom in Excited state  $\uparrow\downarrow \uparrow \uparrow \uparrow$

Now they are 3 half filled orbitals each having one unpaired electron but they belong to different orbitals to get equivalent bonds these 3 orbitals hybridised to form three new equivalent hybrid orbitals.

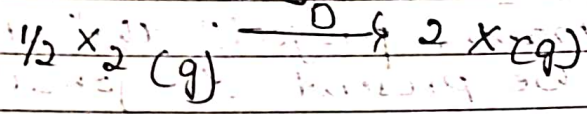
Boron forms  $sp^2$  hybridised orbitals. These 3 orbitals lie in a plane & are directed towards the 3 corners of an equilateral triangle so that the angle between them is  $120^\circ$  as shown in figure.



Step 1: The steps involved in Born-Haber cycle are:  
 Sublimation [S] - Sublimation energy of metal.



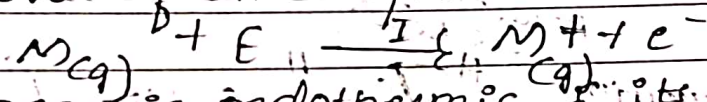
The process is endothermic & its value +ve quantity  
 Step 2: Dissociation [D] - Dissociation energy of  $X_2$  gas atom.



This process is endothermic & its value is +ve quantity.

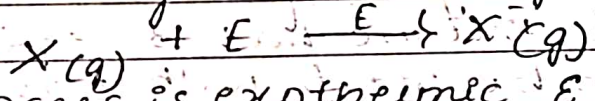
Step 3: Ionisation Energy [I].

Removal of an  $e^-$  from an isolated gaseous atom [M].



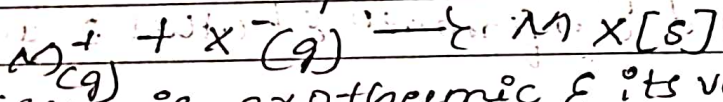
The process is endothermic & its value is +ve quantity.

Step 4: Electron Affinity [E]: The released when a neutral gaseous atom gains.



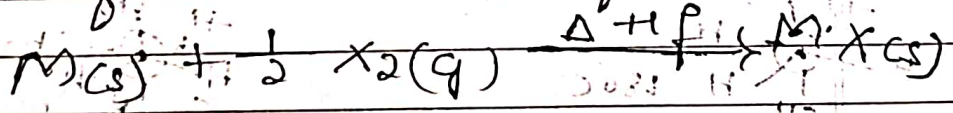
The process is exothermic & its value -ve quantity.

Step 5: Lattice Energy [U] - Energy released when one mole of gaseous +ve & -ve ions.



The process is exothermic & its value is -ve.

Step 6: Heat of Formation [ $\Delta H_f$ ].



$$\Delta H_f = S + \frac{1}{2} D + I + E + U$$

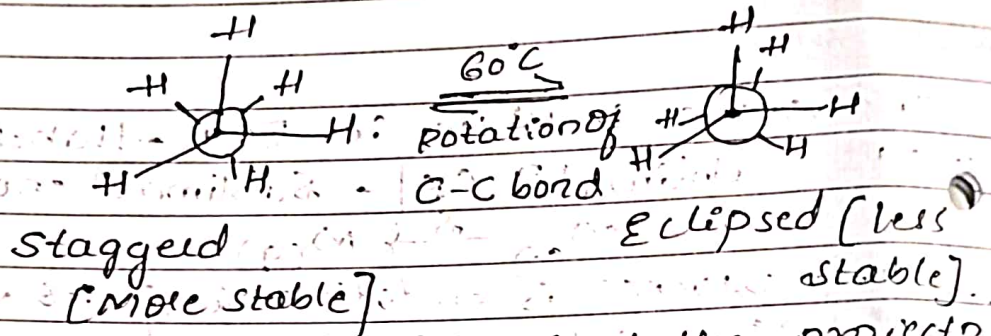
$$U = \Delta H_f - \frac{S + D + I + E}{2}$$

Where,  $U \rightarrow$  Lattice Energy

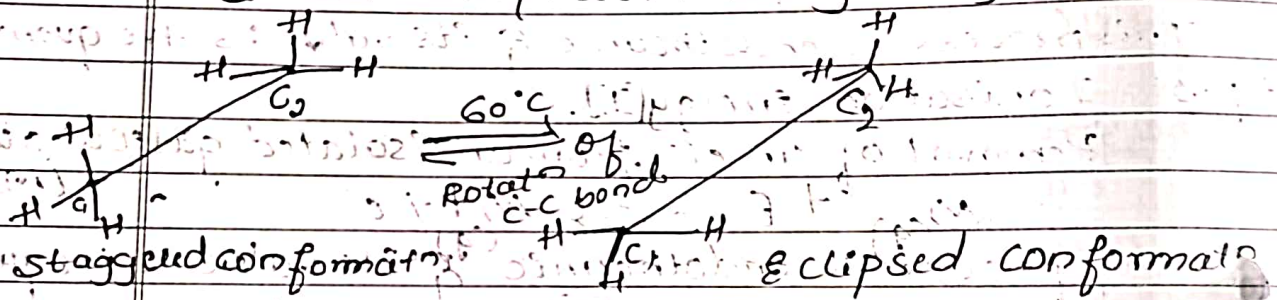
$\Delta H_f \rightarrow$  Heat of formation.



3) Newman Projection: The represent<sup>n</sup> is one in which the 3 dimensional arrangement of a molecule can easily understood by the representat<sup>n</sup> of newproj<sup>n</sup> - ion structure.



b) Sawhorse Projection: In case of the project<sup>n</sup> similar to newman's project<sup>n</sup> the structure of the molecule can be present diagonally the carbon.

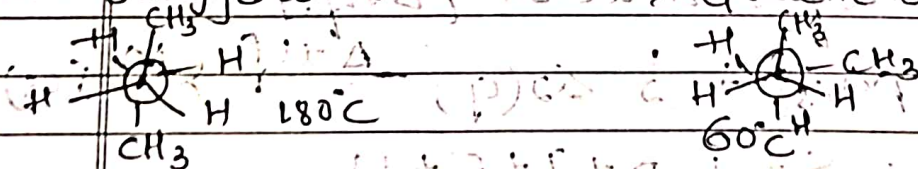


4) Conformational Analysis of n-Butane:

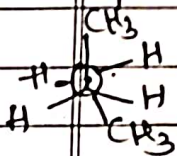
Let us consider butane has 3 C-C single bond a molecule can be rotat with each other.

→ To study the conformation analysis butane molecule which gives 4 possible conformations. 2 are straggled conformat<sup>n</sup> & other 2 are eclipsed confor- mation.

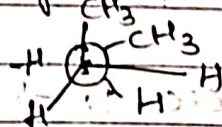
1) Anti straggled 2) Gauche straggled



3) Anti-eclisped



4) Dihedral Eclisped.



Hence it leads high interactn b/w methyl group leads to less stable or highly unstable.



NO I Any three

1) Ionic bond - A bond formed by transfer of electrons b/w atoms is called ionic bond.

2) Lattice energy is the energy that is released when gaseous ions bond to form an ionic solid, or the energy required to break apart an ionic solid into its gaseous ions.

3) Geometric isomers are two or more compounds with the same number & types of atoms & bonds but which have different geometries for the atoms.  
Ex: - Fumaric acid, maleic acid

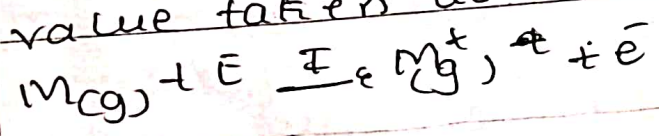
4) The order of a reaction is the relationship b/w the rate of a chemical reaction and the concentration of the reactants.

NO II 2) Born - Haber cycle steps

\* Step 1: - Sublimation: - Sublimation energy of the metal since energy is required for the process its value is taken as a positive quantity

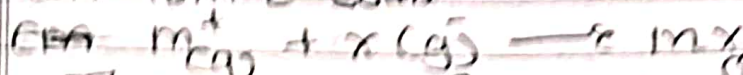
\* Step 2: - Dissociation energy of  $X_2$  gas into atoms  
 $\frac{1}{2} X_2 \rightarrow X$  This process is endothermic its value is taken as +ve quantity

\* Step 3: - Ionisation energy (I): - That is energy required for the removal of an electron from an isolated gaseous atom (M) the process is endothermic and its value taken as +ve quantity



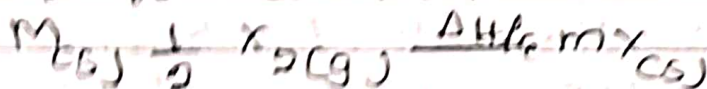
STEP 4. Electron affinity (E): - That is amount of energy released when a neutral gaseous atoms gains an electron the process is Exothermic & its value is taken as a -ve quantity

\* STEP 5: Lattice Energy: - It is amount of energy released when one mole of gaseous positive & negative ion condense to form an ionic solid



The process is Exothermic & its value is negative

\* STEP 6:  $\Delta H_f$ : Heat of formation  
It is defined as the amount of energy released when one mole of a substance is formed from its elements



$$\Delta H_f = S + \frac{1}{2} D + I + E + U$$

$$\therefore U = \Delta H_f - S - \frac{1}{2} D - I - E$$

b) Drop number method

A stalagnometer consist of long glass tube with a bulb in the middle the lower position made up of capillary tube the are 2 marks A & B a rubber tube with a pinch B attached at the upper end with controls the rate of flow

procedure: - 1) The stalagnometer was washed with washing solution & dried by passing hot air a rubber tube with pinch clip is attached at the upper end & stalagnometer is flaped vertically to the stand

STUDENTS NAME		TOTAL MARKS OBTAINED
CLASS	SUBJECT	
ROLL NO	DATE	

ii) The lower end of Stalgnometer is Inversed to the in a beaker containing in liquid & sucked above the mark

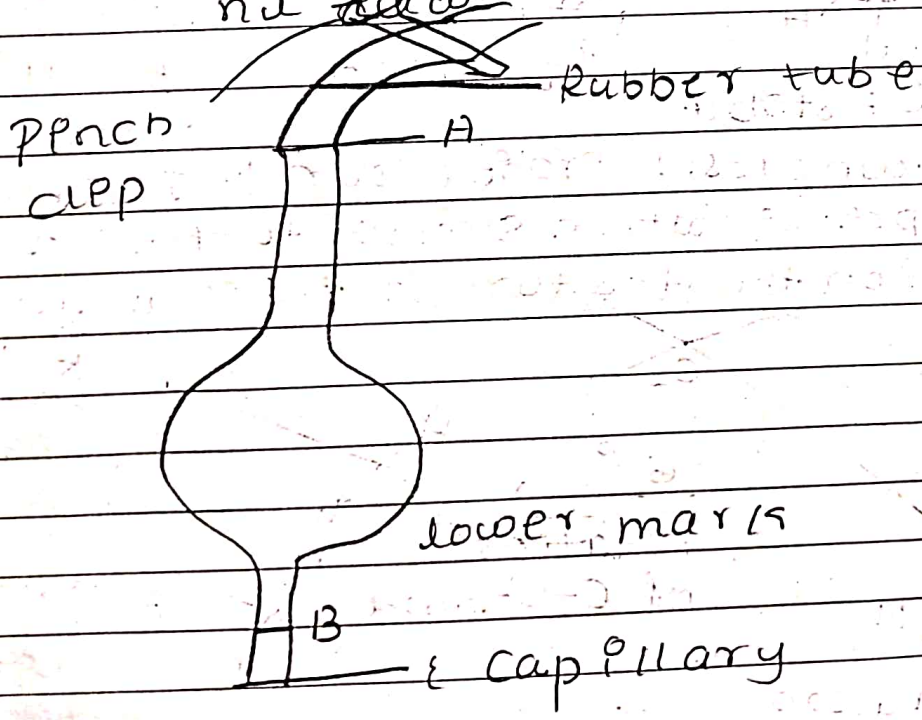
iii) The liquid flows downward the rate of flow of liquid is adjusted so that 18-20 drops per minute do not disturb the adjustment until the counting of drops for all liquids completes

v) liquid is sucked above the mark A & allowed in flow down the counting is started when a level comes to mark A & stop the counting level process lower mark B Note down the number of drops

5) Clean the stalgnometer with washing solution it dry with hot air repeat the produce of liquids B & C & water The density of two liquids are

6) The density of two liquids are determined by using specific gravity bottles calculated the surface tension using The formula

$$V_1 = \frac{m_w \times d_1 \times V_w}{n_d \times d_w}$$

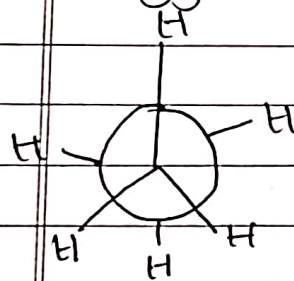


3) Newman projection

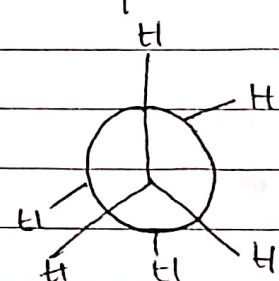
Newman's projection or the representation is one in which the 3-dimensional arrangements of a molecule can be easily understood by the representation of Newman's structure as follows

Let us consider 3 dimensional structure of ethane molecule can be easily represented by the Newman structure. According to Newman's projection of the ethane molecule can be represented into two conformations - the most stable conformation is known as staggered conformation. The other eclipsed conformation will give less stability due to interaction. The interconversion of ethane molecule is as follows

① Staggered conformation      Eclipsed conformation



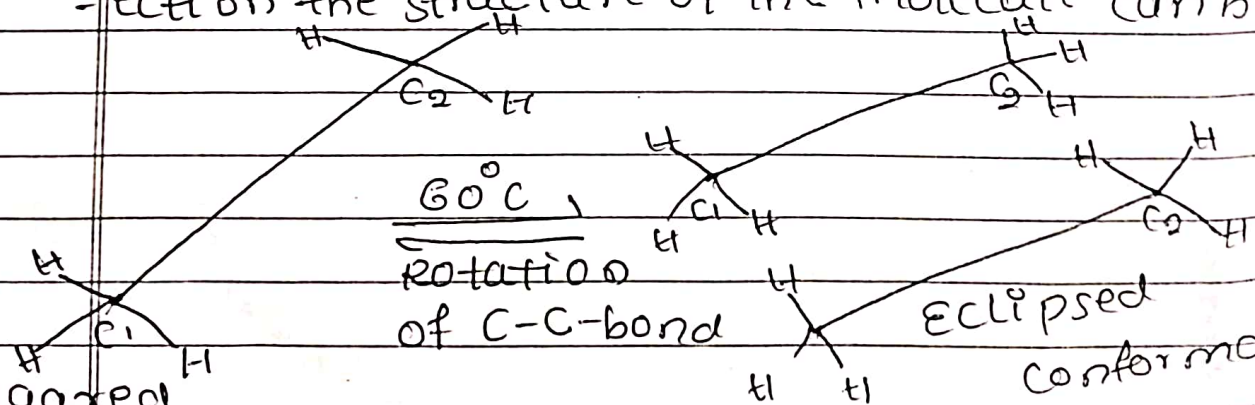
$60^\circ$   
 rotation  
 of C-C bond



- most stable

- less stable

2) Sawhorse projection :- In case of sawhorse representation similar to the Newman's projection the structure of the molecule can be



Staggered conformation

Eclipsed conformation

Sawhorse representation of ethane molecule

B.V.V.Sangha's  
Basaveshwar Science College, Bagalkot  
Department of Chemistry  
Class: B.Sc- II Semester  
List of Slow Learners 2023-2024

**Marks List**

Sl.No	Reg.No	Student name	I Test	II Test
1	U26ZW23S0006	Sachin Irappa Kamashetti	16	18
2	U26ZW23S0029	VISHWANATH ASHOK LAMANI	14	15
3	U26ZW23S0070	PRAVEEN KELUR	13	15
4	U26ZW23S0078	Keerti Shrishail Karigar	12	14
5	U26ZW23S0119	RAMYA GURAPPANAVAR	13	16
6	U26ZW23S0121	SHRIMANT PATIL	14	17
7	U26ZW23S0137	MAHANTES H KALLUKUTTIGAR	15	17
8	U26ZW23S0140	SHREENIDHI HIREMATH	16	17
9	U26ZW23S0164	Pradeep Hadimani	13	15
10	U26ZW23S0167	KRISHNA S BHOSHANAVAR	12	15
11	U26ZW23S0168	MANJUNATH V KELAGINONI	11	14
12	U26ZW23S0176	NOUSHAD MULLA	13	15
13	U26ZW23S0204	SRUSHTI MAREPPA BHOVI	15	17
14	U26ZW23S0215	ANUSHA	16	18
15	U26ZW23S0217	AKSHATA BHIMASHI ARAKERI	18	19
16	U26ZW23S0219	BASAVARAJ RNAGAPPANAVAR	15	17
17	U26ZW23S0254	NIRIKSHA HADIMANI	14	16
18	U26ZW23S0260	CHETAN HUNAGUNDI	15	16
19	U26ZW23S0286	GANGADHAR B SEETIMANI	17	16
20	U26ZW23S0299	POOJA PRAKASH MINAJAGI	13	15
21	U26ZW23S0347	SUNIL SHANKRAPPA HULYAL	14	14
22	U26ZW23S0372	ANJALI SHIVANAND WADAWADAGI	16	15

*[Signature]*  
Dept of  
Chemistry

**Principal**  
B. V. V. S's Basaveshwar  
Science College, Bagalkot

**B.V.V.Sangha's**  
**Basaveshwar Science College. Bagalkote**  
**Department of Chemistry**  
**B.Sc IV semester 2023-2024**  
**Student List ( Slow - Learners )**

SL No	Reg No	Name of the Student	Gender	Category
1	U15JP22S0034	Pallavi Mural	Female	SC
2	U15JP22S0065	Sachin Mounesh Badiger	Male	II A
3	U15JP22S0078	Vikas Chandakavate	Male	II B
4	U15JP22S0100	Sanjeev Gurappa Jali	Male	GM
5	U15JP22S0101	Sabanaaz Solapur	Female	GM
6	U15JP22S0108	Manjunath Yankanchi	Male	GM
7	U15JP22S0142	Vinay S Havalaramata	Male	III A
8	U15JP22S0156	Hucchesh Y Majjagi	Male	II A
9	U15JP22S0160	Shrishail Patil	Male	II A
10	U15JP22S0169	Priyanka T Aramani	Female	GM
11	U15JP22S0181	Shankamma M Hadimani	Female	II A
12	U15JP22S0208	Tejashwini Badiger	Female	II A
13	U15JP22S0214	Suprita	Female	ST
14	U15JP22S0257	Neviedita Balavantanavar	Female	II A
15	U15JP22S0259	Sangamesh Bhajantri	Male	SC
16	U15JP22S0328	Chetan Kamat	Male	SC
17	U15JP22S0384	Ashraf Javedsab Valikar	Female	II B


  
H.O.D.  
( Smt S.S. Karishankari )

  
**Principal**  
B. V. V. S' s Basaveshwar  
Science Collage, Bagalkot

**B.V.V.Sangha's**  
**Basaveshwar Science College, Bagalkote**  
**Department of Chemistry**  
**B.Sc VI semester 2023-2024**  
**Student List ( Slow - Learners )**

SL No	Reg No	Name of the Student	Gender	Category
1	U15JP21S0053	Beerappa Prabhu Bennura	Male	II A
2	U15JP21S0076	Adivappa Pattanad	Male	III B
3	U15JP21S0085	Akash Hanamappa Pujar	Male	III B
4	U15JP21S0214	Kalmesh G Tumbaramatti	Male	II A
5	U15JP21S0306	Nagaraja B Vaddar	Male	SC
6	U15JP21S0316	Panchaksharayya K Salimath	Male	III B
7	U15JP21S0323	Poornima S Mugad	Female	III B
8	U15JP21S0352	Sampatkumar Biradar	Male	IIA
9	U15JP21S0374	Sangamesh Naganoor	Male	III A
10	U15JP21S0406	Siddaram Vittal Kaladagi	Male	II A
11	U15JP21S0420	Sunilkumar B Majjagi	Male	II A
12	U15JP21S0448	Vinayak G Benalamath	Male	GM
13	U15JP21S0450	Vinayak Vittal Ambiger	Male	C-I
14	U15JP21S0453	Vishwanath S Chikkalaki	Male	II A


  
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( Smt S.S.Karishankari )

  
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**Basaveshwar Science College, Bagalkote**  
**Department of Mathematics**  
**Slow Learners List:2023-24**  
**Class: B.Sc II sem**

Sl. No	Reg No	Name of the student
1	U26ZW23S0022	ABHISHEK BIRADAR
2	U26ZW23S0030	GOURAMMA MASUTI
3	U26ZW23S0055	PRIYANKA LAMANI
4	U26ZW23S0073	NIRMALA JOGI
5	U26ZW23S0159	ANJALI LAMANI
6	U26ZW23S0176	NOUSHAD MULLA
7	U26ZW23S0213	PAVAN GARAWAD
8	U26ZW23S0440	POOJA BASAPPA PATIL
9	U26ZW23S0021	Chaitanya Popat Magdum
10	U26ZW23S0081	MALLIKARJUN
11	U26ZW23S0127	RAJESHWARI S
12	U26ZW23S0141	RASHMI SALAPUR
13	U26ZW23S0161	UMARFAROOQ KAKHANDAKI
14	U26ZW23S0456	PAVANKUMAR KATTIMANI
15	U26ZW23S0282	PRIYANKA HUNDEKAR



H.O.D.

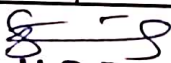
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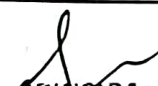
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**Basaveshwar Science College, Bagalkot**  
**Department of Mathematics**  
**Year 2023-24**  
**Class: B.Sc-IV semester**  
**List of slow learner students**

Sl.No	Reg.No	Name of the Students	DSC	Gender	Caste
1	U15JP22S0007	Sanjana Goudar	CsM	F	IIIA
2	U15JP22S0024	Ashwini Basarakod	PM	F	IIA
3	U15JP22S0066	Darshan Goudar	MP	M	ST
4	U15JP22S0084	Anil Kumbar	CM	M	IIA
5	U15JP22S0092	Ajayakumar Nayak	CsM	M	ST
6	U15JP22S0098	AmreenNaaz Kolhar	MP	F	IIB
7	U15JP22S0105	Abhishek Malagond	PM	M	IIA
8	U15JP22S0111	Ramanagouda Patil	PM	M	IIA
9	U15JP22S0114	Kashinath Hulagannavar	MC	M	IIA
10	U15JP22S0156	Hucchesh Majjagi	MC	M	IIA
11	U15JP22S0250	Sagar Goudar	PM	M	IIA
12	U15JP22S0252	Vidyashree Tumbagi	PM	F	IIIA
13	U15JP22S0262	Laxman Madar	MP	M	SC
14	U15JP22S0266	Sunil Karigar	PM	M	IIA
15	U15JP22S0286	Hanamanth Bajantri	PM	M	SC
16	U15JP22S0303	Ankitsingh Thakur	CsM	M	IIA
17	U15JP22S0339	Pundleek Poojari	MP	M	IIIB
18	U15JP22S0344	Siddartha Sobanad	PM	M	IIIB
19	U15JP22S0345	Bhimanagouda Nainapur	PM	M	IIA
20	U15JP22S0377	Priyanka Madar	PM	F	SC

  
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**Department of Mathematics**  
**Slow Learners List: 2023-24**  
**B.Sc VI Semester**

Sl. No.	Reg. No.	Name of the Student
1	U15JP21S0009	AKSHATA HIREMATH
2	U15JP21S0111	CHANDRASHEKHAR S
3	U15JP21S0151	DARSHAN N SHIVAPUJIMATH
4	U15JP21S0286	PRIYANKA S INDRI
5	U15JP21S0324	RAHUL BHEEMAPPA GANIGER
6	U15JP21S0362	SANDESH D BANDIWADDAR
7	U15JP21S0367	SANJAYAKUMAR
8	U15JP21S0389	SHASHIDHAR ONI
9	U15JP21S0412	SUDEEP SURESH RATHOD
10	U15JP21S0414	SUSHMA I HAVARAGI
11	U15JP21S0425	UMESH P MADAR
12	U15JP21S0435	VARSHITA R DAPALAPUR
13	U15JP21S0454	VIJAY PRAKASH GARI



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**Department of Botany**

**Students List of Slow Learners: 2023-24**

**B.Sc-I & II semester**

Sl. No.	Name of the Students	Register No.
1.	SACHIN I KAMASHETTI	U26ZW23S0006
2.	BHAGYASHREE S BIRADAR	U26ZW23S0011
3.	ADITI RAJU KALYANI	U26ZW23S0017
4.	SAVITRI	U26ZW23S0018
5.	KRISHNA K HOSAMANI	U26ZW23S0033
6.	ARCHANA B SADDI	U26ZW23S0039
7.	BHARMANNA	U26ZW23S0043
8.	SHRAVAN GALAGALI	U26ZW23S0066
9.	KEERTI SHRISHAIL KARIGAR	U26ZW23S0078
10.	MAHANANDA MALLAYYA HIREMATH	U26ZW23S0080
11.	DARSHAN	U26ZW23S0118
12.	RAMYA GURAPPANAVAR	U26ZW23S0119
13.	VENKATESH NAYAK	U26ZW23S0132
14.	VIJAYASHREE GADDEPPA PUJARI	U26ZW23S0133
15.	YAMUNA MALLAPPA NYAMAGOND	U26ZW23S0135



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**Department of Botany**

**Students List of Slow Learners: 2023-24**

**B.Sc-III & IV semester**

<b>Sl. No.</b>	<b>Name of the Students</b>	<b>Register No.</b>
1.	Basavaraj Desai	UI5JP22S0210
2.	Chetan Kamat	UI5JP22S0328
3.	Gouramma P	UI5JP22S0347
4.	Jyoti Nilugal	UI5JP22S0361
5.	Madanakumar Hosamani	UI5JP22S0298
6.	Mallikarjun	UI5JP22S0329
7.	Pooja Hakati	UI5JP22S0349
8.	Pooja Naragund	UI5JP22S0341
9.	Prajwal Chavan	UI5JP22S0318
10.	Sanjeev Jali	UI5JP22S0100
11.	Shankamma Hadimani	UI5JP22S0181
12.	Veeresh Maktali	UI5JP22S0363
13.	Priyanka t Aramani	UI5JP22S0169
14.	Ravi Kambar	UI5JP22S018

  
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**Department of Botany**

**Students List of Slow Learners: 2023-24**

**B.Sc-V & VI semester**

Sl. No.	Name of the Students	Register No.
1.	Akshata P Kalagi	U15JP21S0010
2.	Anush S Kengal	U15JP21S0020
3.	Bhagya Chandappa	U15JP21S0055
4.	Afreen Chandanakeri	U15JP21S0077
5.	Harish S Kiraur	U15JP21S0192
6.	Kalamesh G Tumbaramatti	U15JP21S0214
7.	Karthik Shivappa Nyamagoudar	U15JP21S0225
8.	Megha Yallappa Gaddanakeri	U15JP21S0268
9.	Manoj B Paranagoudar	U15JP21S0279
10.	Praveen S Inganal	U15JP21S0338
11.	Sampatakumar Biradar	U15JP21S0352
12.	Shivanand M Jokali	U15JP21S0387

  
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**Department of Biotechnology**  
**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: I SEM**


S.No	Student Reg .No	Name
01	U15jp2250297	Anjali Shivanand Wadawadagi
02	U15jp2250329	Basavaraj Nagappa Bati
03	U15jp2250150	Krishna Kashinath Hosamani
04	U15jp2250157	Nandish Siddappa Choteppagol
05	U15jp2250331	Pramodraj Gaded
06	U26zw23s0426	Rohan Naik
07	U26zw23s0322	Savita Madabhavi

**Basaveshwar Science College, Bagalkot**  
**Department of Biotechnology**  
**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: III SEM**

S.No	Student Reg .No	Name
01	U15jp2250012	Laxmi
02	U15jp2250297	Ameeth Haddapad
03	U15jp2250298	Madanakumar Hosamani
04	U15jp2250329	Mallikarjun
05	U15jp2250150	Shivaleela

  
Head  
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**Department of Biotechnology**

**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: II SEM**

S.No	Student Reg .No	Name
01	U15jp2250297	Anjali Shivanand Wadawadagi
02	U15jp2250329	Basavaraj Nagappa Bati
03	U15jp2250150	Krishna Kashinath Hosamani
04	U15jp2250157	Nandish Siddappa Choteppagol
05	U15jp2250331	Pramodraj Gaded
06	U26zw23s0426	Rohan Naik
07	U26zw23s0322	Savita Madabhavi

**Basaveshwar Science College, Bagalkot**

**Department of Biotechnology**

**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: IV SEM**

S.No	Student Reg .No	Name
01	U15jp2250012	Laxmi
02	U15jp2250297	Ameeth Haddapad
03	U15jp2250298	Madanakumar Hosamani
04	U15jp2250329	Mallikarjun
05	U15jp2250150	Shivaleela

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**Department of Biotechnology**

**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: V SEM**

<b>S.No</b>	<b>Name of the student</b>	<b>Register no</b>
01	Abhishek davangeri	U15JP21S0004
02	Adivappa.pattanad	U15JP21S0076
03	Giriyappa. h	U15JP21S0075
04	Ningaraj.shirur	U15JP21S0311
05	Pooja.pawar	U15JP21S0319
06	Pratibha.bajantri	U15JP21S0329
07	Shreedar.b	U15JP21S0396



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**Department of Biotechnology**


**List of Slow Learner 2023-24**

**Class: B.Sc**

**Semester: VI SEM**

<b>S.No</b>	<b>Name of the student</b>	<b>Register no</b>
01	Abhishek davangeri	U15JP21S0004
02	Adiveppa.pattanad	U15JP21S0076
03	Giriyappa. h	U15JP21S0075
04	Ningaraj.shirur	U15JP21S0311
05	Pooja.pawar	U15JP21S0319
06	Pratibha.bajantri	U15JP21S0329
07	Shreedar.b	U15JP21S0396

  
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**BASAVESHWAR SCIENCE COLLEGE, BAGALKOT**  
**DEPARTMENT OF MICROBIOLOGY**

Class: B.Sc-I year  
2023-24

**List of Slow learner Students**

Sl.No	Exam No	Name of the Students
1	U26ZW23S0045	SANIYA MUDDEBIHAL
2	U26ZW23S0137	MAHANTESH H K
3	U26ZW23S0143	RAJU P M
4	U26ZW23S0167	KRISHNA S B
5	U26ZW23S0261	MEGHANANDA M
6	U26ZW23S0283	AISHWARYA M J
7	U26ZW23S0306	MUTTU INGALESHWAR
8	U26ZW23S0377	HANAMANT R NAYKAR
9	U26ZW23S0445	MALLIKARJUN S ROLLI

  
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**Department of Microbiology**  
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DEPARTMENT OF MICROBIOLOGY

Class: B.Sc-II year

2023-24

**List of Slow learner Students**

Sl.No	Exam No	Name of the Students
1	U15JP22S0035	ANITA VENKATESH PUDUR
2	U15JP22S0042	ANNAPOORNA G KADAGAD
3	U15JP22S0142	VINAY .S. HAVALDARAMATA
4	U15JP22S0371	SRUSHTI BHAGAVATI
5	U15JP22S0384	ASHRAF JAVEDSAB VALIKAR
6	U15JP22S0390	DODDABASAYYA N H
7	U15JP22S0391	HANAMANTH B.T

  
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**Department of Microbiology**  
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**BASAVESHWAR SCIENCE COLLEGE, BAGALKOT**  
**DEPARTMENT OF MICROBIOLOGY**

Class: B.Sc-III year  
2023-24

**List of Slow learner Students**

Sl. No	Exam No	Name of the Students
1	U15JP21SO146	HARISH R BHAJANTRI
2	U15JP21SO248	MANJUNATH S HALEMANI
3	U15JP21SO284	MITHUN P LAMANI
4	U15JP21SO308	NAVEEN S CALAWADI
5	U15JP21SO350	SAMEERA M JARDI
6	U15JP21SO374	SANGAMESH M NAGNOOR
7	U15JP21SO3i4	NIVEDITA GANGAIR
8	U15JP21SO469	VIJAYAKUMAR B PARAPPANAVAR
9	U15JP21SO450	VINAYAK V AMBIGER

  
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**Department of Microbiology**  
**Basaveshwar Science College, Bagalkot**

  
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